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**Practical/Laboratory Manual Chemistry Class - XI - Er. Meera Goyal 2021-05-29**

**1.Basic Laboratory Techniques** 1.To cut a glass tube or glass rod, 2.To bend the glass rod at an angle, 3.To draw a glass jet from a glass tube 4.To bore a cork and fit a glass tube into it Viva-Voce 2.Characterisation and Purification of Chemical Substances 1.To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique) Viva-Voce 2.To determine the boiling point of a given liquid when available in small quantity (simple laboratory method) Viva-Voce 3.To prepare crystals of pure potash alum  $[K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O]$  from the given impure sample 4.To prepare the pure crystals of copper sulphate from the given crude sample 5.To prepare pure crystals of benzoic acid from a

given impure sample Viva-Voce 3.Measurement of pH Values 1.To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper 2.To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH<sub>3</sub>COOH) of same concentration 3.To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper 4.To study the pH change by common ion (CH<sub>3</sub>COO<sup>-</sup> ion) in case of weak acid (CH<sub>3</sub>COOH) 5.To determine the change in pH value of weak base (NH<sub>4</sub>OH) in presence of a common ion (NH<sub>4</sub><sup>+</sup>) Viva-Voce 4.Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2.To study the shift in equilibrium between  $[Co(H_2O)_6]^{2+}$  and Cl<sup>-</sup> ions by changing the concentrations of either of the ions Viva-Voce 5. Quantitative Analysis 1.To prepare M/10 oxalic acid solution by direct weighing method 2.To prepare M/10 solution of sodium carbonate by direct weighing method 3.To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid 4.To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution Viva-Voce 6.Qualitative Analysis Analysis of Anions Analysis of Cations Viva-Voce 7.Detection of Elements in Organic Compounds 1.To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test 2.To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number ..... by Lassaigne's test Viva-Voce INVESTIGATORY PROJECTS 1.Checking of Bacterial Contamination in Water 1.To check the bacterial contamination in drinking water by testing sulphide ions Viva-Voce 2. Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation 2. To purify water by boiling 3. o purify water by distillation method 4. To purify water by reverse osmosis technique 5. To purify water by GAC method 6. To purify water by bleach treatment 7. To purify water by oxidising agent 8. To purify water by ozone treatment method Viva-Voce 3. Water Analysis 1.To test the hardness of different water samples Viva-Voce 4. Foaming Capacity of Various Soaps 1 .To compare the foaming capacity of different washing soaps 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva-Voce 5. Tea Analysis 1.To study the acidity of different samples of tea leaves (tea) by using pH paper Viva-Voce 6.Analysis of Fruits and Vegetable Juices 1. To analysis the fruit and vegetable juices for the constituent present in them Viva-Voce 7. Rate of Evaporation 1. To study the rate of evaporation of different liquids IViva-Voce 8. Effect of Acids and Bases on Tensile Strength of Fibres 1.To compare the tensile strength of natural fibres and synthetic fibres 2.To study the effect of acids and bases on tensile strength of different fibres Viva-Voce

**Oswaal CBSE Chapterwise & Topicwise Question Bank Class 12 Chemistry Book (For 2023-24 Exam) - Oswaal Editorial Board 2023-01-09**

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**Chemistry Lab Manual - Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar Lab Manual**

**Comprehensive Experimental Chemistry - V. K. Ahluwalia 1997**

This Book Has Been Especially Written For Class Xii Students Under 10+2 Pattern Of Education According To The Syllabi Prescribed By The Cbse And Other States Boards. This Book Will Help The Students In Acquiring Correct Skills In Practicals And Various Techniques Of All Laboratory Experiments.Salient Features \* An Introduction To The Book Is Given. This Describes The Laboratory Apparatus And Instructions And Precautions For Working In The Laboratory. \* Simple Language And Lucid Style. \* Adequate Number Of Illustrations To Explain And To Clarify The Use Of Various Apparatus Used In The Laboratory. \* Theoretical Aspects Of Each Equipment Have

Been Discussed Along With Experiments. \* In Volumetric Analysis, Both The Normality And Molarity Concepts Are Made Clear. \* Li>In Quantitative Analysis (Inorganic And Organic), Various Tests Have Been Given In A Systematic Way. Specimen Recordings Of Experiments Are Given To Help The Students To Record On Their Notebooks. \* Viva-Voice Questions Have Been Included In Each Chapter. \* A Fairly Large Number Of Investigatory Projects Covering Various Topics Are Given. Selection Of Projects Is Carefully Made Which Can Be Easily Performed In School Laboratory. \* An Appendix Describing Various Chemical Hobbies Is Given Which Will Be Extremely Helpful To The Students For The Development Of Chemical Hobbies, Understanding The Basic Principles Involved And The Chemistry Of Various Hobbies. \* An Appendix Describing Some Typical Chemical Exhibits Is Also Given. This Will Help The Students To Participate In The Science Fairs Organized By Various Agencies. These Experiments Will Cultivate Interest Among The Students For Learning Chemistry. \* An Appendix Each For The Solubility'S Of Various Salts, Atomic Weights, Preparation Of Various Reagents, Indicator Papers And The First Aid To Be Administered In Case Of Accidents Is Given. The Syllabi Prescribed For Class Xii Students Under 10+2 Pattern Along With Distribution Of Marks Is Also Given.

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*Methods Of Teaching Science - K.Jaya Sree 2010*

The method of teaching each subject play a pivotal role in enhancing the efficiency of their practitioners. Identifying the very importance of the methods of teaching and the quality of books, a series of books on the methods of teaching different subjects have been developed by experienced teacher educators for the benefit of teachers in making in teacher education institutions. Contents: Teacher s Role, Teaching Techniques, Methods of Vogue, Approaches in Vogue, Aims and Objectives of Teaching, Advancement of Science in India, Behaviour and Objectives, Educational Technology, Audio-visual Aids in Use, Experiments in Innovation, Programmes for Enrichment, Instruction in a Programmed Manner, Individual Level Instructions, Planning the Lessons, Curriculum (India), Curriculum (World), Textbook and Material Projects, Social Service.

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Lab Manuals

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Physics, Chemistry & Maths 2022-2023 is designed to offer a better understanding of the topics and concepts to score maximum in ISC class 12 board exams 2023. Students are required to get this ISC Class 12 sample Paper for English Paper-1, English Paper-2, Physics, Chemistry & Maths 2022-2023 to boost their confidence about a particular topic or the entire chapter according to their needs. It is to assist in understanding the board examination scheme and clarity of concepts for exam preparations.

Chemistry - R P Manchanda

A text book on Chemistry

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Investigatory Projects in Chemistry - A. K. Chakravarty 1983-01

*Teaching Chemistry Around the World* - Björn Risch 2010

As teachers we often tend to expect other countries to teach chemistry in much the same way as we do, but educational systems differ widely. At Bielefeld University we started a project to analyse the approach to chemical education in different countries from all over the world: Teaching Chemistry around the World. 25 countries have participated in the project. The resulting country studies are presented in this book. This book may be seen as a contribution to make the structure of chemistry teaching in numerous countries more transparent and to facilitate communication between these countries. Especially in the case of the school subject chemistry, which is very unpopular on the one hand and occupies an exceptional position on the other hand - due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving - we have to learn from each others' educational systems.

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Investigatory Projects in Chemistry - Pradip Tekade 2011-07

In recent years, the courses in chemistry have been considered extremely important for students desirous of pursuing basic science as well as technical education. Taking into consideration this trend, this book has been specially written for students who are interested in investigatory and innovative projects in chemistry. The unique feature of this book is that the basic, theory, procedure and conclusions for each project are given in comprehensive manner. In all 51 projects are included in this book from all the branches of chemistry viz. inorganic, organic, physical, analytical and general chemistry. Some of them are: Antacid effectiveness analysis, Study of tea chemistry, study the foaming capacity of soaps, Study of constituents of brass and bronze, Preparation of potash alum from scrap aluminum, Extraction of essential oil from aniseed, preparation of pigments and poster paints using chemicals and reagents, Analysis of calcium, phosphate, chloride, magnesium and iron from bone ash, To Study the Setting of Cement, Comparative study and qualitative analysis of cold drinks, preparation of cuprammonium rayon threads from filter paper, Analysis of talcum powder etc.

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